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APPLIED CHEMISTRY I - MAIT4us

VIVA QUESTIONS - PRACTICALS 1. DETERMINATION OF HARDNESS OF WATER. 1. What is temporary hardness? Hardness of water due to presence of carbonates and bicarbonates of Calcium and magnesium is

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Model Viva Questions Chemistry Lab Common to : CS, IT, ET 1st Sem Title of the practical: Identification of two cations and two anions in a sample of mixture. Q1: What is qualitative analysis? A1: The type of analysis that deals with the methods which are used to determine the constituents of a compound.

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CHEMISTRY PRACTICAL VIVA QUESTIONS AND ANSWERS

Chemistry Lab(15CHEL17/27) Viva Questions – 1: INSTRUMENTAL EXPERIMENTS: 1. Estimation of FAS potentiometrically using standard $K_2Cr_2O_7$ solution. 2. Estimation of Copper colorimetrically. 3. Estimation of Acids in acid mixture conductometrically. 4. Determination of pKa of weak acid using pH meter. 5.

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Organic Chemistry Questions The Covalent Bond 1. The hybridization of the central carbon in CH_3CN and the bond angle CCN are a. sp^2 , 180° . b. sp , 180° . c. sp^2 , 120° . d. sp^3 , 109° . 2. Which of the following statements about an sp hybridized carbon is FALSE?

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11. Students should update the records and lab manuals session wise. Before leaving the lab the student should get his lab manual signed by the faculty. 12. Students should not move around in the lab. 13. If any emergency, the student should take the permission from faculty member concerned in written format. 14.

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method. 3. Determination of percentage of Copper in brass using standard sodium thiosulphate solution. 4.

Chemistry Lab (15CHEL17/27) Viva Questions-2

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Chemistry Lab Common to: CS, IT, ET 1 st Sem Title of the practical: Identification of two cations and two anions in a sample of mixture.

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(Pr) 6.3 Chemistry Lab CH (Pr) 6.4 Chemistry Lab Viva questions may be asked on any of the experiments prescribed Valence bond Theory of coordination compounds with reference to $(Fe(CN)_6)_4^-$, Question number 1-12 carries 2marks-answer any 10 questions-marks:20.

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1. Question No. 1 should be compulsory and cover the entire syllabus. This question should have objective or short answer type questions. It should be of 25 marks. 2. Every unit should have two questions. However, student may be asked to attempt only 1 question from each unit. Each question should be of 12.5 marks.

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